

PHYSICS 140A : STATISTICAL PHYSICS
HW ASSIGNMENT #6

(1) $\nu = 8$ moles of a diatomic ideal gas are subjected to a cyclic quasistatic process, the thermodynamic path for which is an ellipse in the (V, p) plane. The center of the ellipse lies at $(V_0, p_0) = (0.25 \text{ m}^3, 1.0 \text{ bar})$. The semimajor axes of the ellipse are $\Delta V = 0.10 \text{ m}^3$ and $\Delta p = 0.20 \text{ bar}$.

- (a) What is the temperature at $(V, p) = (V_0 + \Delta V, p_0)$?
- (b) Compute the net work per cycle done by the gas.
- (c) Compute the internal energy difference $E(V_0 - \Delta V, p_0) - E(V_0, p_0 - \Delta p)$.
- (d) Compute the heat Q absorbed by the gas along the upper half of the cycle.

(2) Determine which of the following differentials are exact and which are inexact.

- (a) $xy \, dx + xy \, dy$
- (b) $(x + y^{-1}) \, dx - xy^{-2} \, dy$
- (c) $xy^3 \, dx + 3x^2y^2 \, dy$
- (d) $(\ln y + \ln z) \, dx + xy^{-1} \, dy + xz^{-1} \, dz$

(3) Liquid mercury at atmospheric pressure and temperature $T = 0^\circ \text{ C}$ has a molar volume of $14.72 \text{ cm}^3/\text{mol}$ and a specific heat at constant pressure of $c_p = 28.0 \text{ J/mol}\cdot\text{K}$. Its coefficient of expansion is $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_p = 1.81 \times 10^{-4}/\text{K}$ and its isothermal compressibility is $\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_T = 3.88 \times 10^{-12} \text{ cm}^2/\text{dyn}$. Find its specific heat at constant volume c_V and the ratio $\gamma = c_p/c_V$. [Reif problem 5.10]

(4) ν moles of an ideal diatomic gas are driven along the cycle depicted in Fig. 1. Section AB is an adiabatic free expansion; section BC is an isotherm at temperature $T_A = T_B = T_C$; CD is an isobar, and DA is an isochore. The volume at B is given by $V_B = (1 - x)V_A + xV_C$, where $0 \leq x \leq 1$.

- (a) Find an expression for the total work W_{cycle} in terms of ν, T_A, V_A, V_C , and x .
- (b) Suppose $V_A = 1.0 \text{ L}$, $V_C = 5.0 \text{ L}$, $T_A = 500 \text{ K}$, and $\nu = 5$. What is the volume V_B such that $W_{\text{cycle}} = 0$?

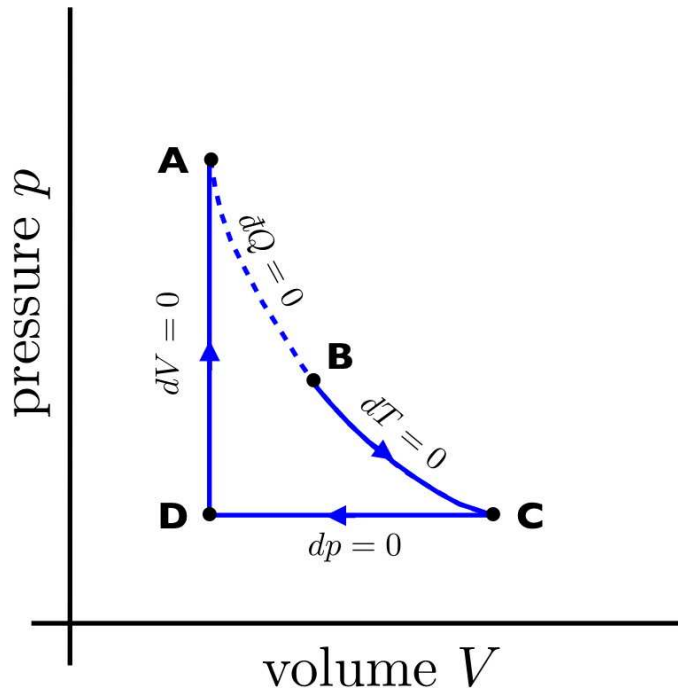


Figure 1: Thermodynamic cycle for problem 4, consisting of adiabatic free expansion (AB), isotherm (BC), isobar (CD), and isochore (DA).

(5) A strange material found stuck to the bottom of a seat in Warren Lecture Hall 2001 obeys the thermodynamic relation $E(S, V, N) = a S^6 / V^2 N^3$, where a is a dimensionful constant.

- (a) What are the MKS dimensions of a ?
- (b) Find the equation of state relating p , V , N , and T .
- (c) Find the coefficient of thermal expansion $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_p$. Express your answer in terms of intensive quantities p , T , and $n = N/V$.
- (d) Find the isothermal compressibility $\kappa = -\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_T$. Express your answer in terms of intensive quantities p , T , and $n = N/V$.