

Problem 1

Classically,  $M = m_1 + m_2$ . Momentum conservation gives:

$$m_1 u - m_2 u = M u' \Rightarrow \boxed{u' = \frac{m_1 - m_2}{m_1 + m_2} u} \quad (a)$$

(b) Relativistically:

Momentum conservation:  $\gamma m_1 u - \gamma m_2 u = \gamma' M u'$

Energy conservation:  $\gamma m_1 c^2 + \gamma m_2 c^2 = \gamma' M c^2 \Rightarrow$

$$\Rightarrow \gamma' M = \gamma (m_1 + m_2) \Rightarrow$$

Substituting in momentum conservation eq.

$$\gamma (m_1 - m_2) u = \gamma' M u' = \gamma (m_1 + m_2) u'$$

$$\Rightarrow \boxed{u' = \frac{m_1 - m_2}{m_1 + m_2} u} \quad (b)$$

it is same as value found classically.

(c) From energy equation,

$$\boxed{M = \frac{\gamma}{\gamma'} (m_1 + m_2)}$$

$$\gamma = \frac{1}{\sqrt{1 - u^2/c^2}}, \quad \gamma' = \frac{1}{\sqrt{1 - u'^2/c^2}} \quad ; \quad \text{since } u' < u \Rightarrow \gamma' < \gamma \Rightarrow$$

$$\Rightarrow \frac{\gamma}{\gamma'} > 1 \Rightarrow \boxed{M > m_1 + m_2}$$

because in an inelastic collision, kinetic energy gets transformed into mass.

## Problem 2

$$(a) \quad \lambda_m T = \frac{hc}{4.96 k_B} \Rightarrow \lambda_0 = \frac{12,400 \times 11,600 \text{ \AA}}{4.96 \times 5000} = 5800 \text{ \AA}$$

$$\boxed{\lambda_0 = 5800 \text{ \AA}}$$

(b) if  $T = 5000^\circ\text{K}$ ,  $T' = 10,000^\circ\text{K} = 2T$ , we have:

$$\frac{hc}{\lambda_0 k_B T} = 4.96, \quad \frac{hc}{\lambda_0 k_B T'} = \frac{4.96}{2} = 2.48$$

Power is proportional to  $\frac{1}{e^{\frac{hc}{\lambda k_B T}} - 1}$ ; so

$$\frac{\text{power}(T')}{\text{power}(T)} = \frac{e^{4.96} - 1}{e^{2.48} - 1} = \frac{141.59}{10.94} = 12.94$$

So power increases by factor 12.94

(c) Total power is proportional to  $T^4$  (Stefan law)

$T$  increases by factor of 2  $\Rightarrow$  total power increases by factor 16

(d) At very large  $\lambda$

$$\frac{1}{e^{\frac{hc}{\lambda k_B T}} - 1} \approx \frac{1}{1 + \frac{hc}{\lambda k_B T} - 1} = \frac{\lambda k_B T}{hc} \quad (\text{using } e^x \approx 1+x)$$

so power is proportional to  $T \Rightarrow$

power increases by factor of 2 at wavelength  $\lambda \gg \lambda_0$

### Problem 3

The ions are in their ground state, so  $n=1$  for the electron.

$E_1 = -E_0 Z^2$  with  $E_0 = 13.6 \text{ eV}$ . To ionize, photon has to have

$$\text{energy } hf = \frac{hc}{\lambda} = E_0 Z^2 \Rightarrow \lambda = \frac{hc}{E_0 Z^2} = \frac{12,400 \text{ eV}\text{\AA}}{13.6 \text{ eV} \cdot Z^2}$$

$$\Rightarrow \lambda = \frac{911.76 \text{ \AA}}{Z^2}. \text{ So to ionize atoms with } Z=1, 2, 3, 4, \dots$$

requires  $\lambda = 911 \text{ \AA}, 228 \text{ \AA}, 101 \text{ \AA}, 57 \text{ \AA}, \dots$

Here, radiation is in range  $100 \text{ \AA}$  to  $150 \text{ \AA} \Rightarrow \boxed{Z=3}$  (a)

The highest energy photon has energy  $hf = \frac{hc}{\lambda}$  with  $\lambda = 100 \text{ \AA} \Rightarrow$

$$\boxed{hf = 124 \text{ eV}}. \text{ The ionization energy is } E_0 Z^2 = 13.6 \text{ eV} \times 9 = \boxed{122.4 \text{ eV}}$$

$$\Rightarrow \text{maximum kinetic energy of electron} = 124 \text{ eV} - 122.4 \text{ eV} = \boxed{1.6 \text{ eV}} \text{ (b)}$$

(c) Wavelengths required for transitions from  $n=1$  to  $m > 1$ :

$$\frac{hc}{\lambda} = Z^2 E_0 \cdot \left(1 - \frac{1}{m^2}\right) \Rightarrow \lambda_{1m} = \frac{hc}{Z^2 E_0} \frac{1}{1 - \frac{1}{m^2}} = \frac{101.307 \text{ \AA}}{1 - \frac{1}{m^2}}$$

$$\Rightarrow \lambda_{12} = \frac{101.307 \text{ \AA}}{\frac{3}{4}} = 135.08 \text{ \AA}$$

$$\lambda_{13} = \frac{101.307 \text{ \AA}}{\frac{8}{9}} = 113.97 \text{ \AA}$$

$$\lambda_{14} = \frac{101.307 \text{ \AA}}{\frac{15}{16}} = 108.06 \text{ \AA}$$

$$\lambda_{1m} < \lambda_{14} \text{ for } m > 4$$

2 absorption lines are seen if incident wavelengths are in range  $110 \text{ \AA}$  to  $150 \text{ \AA}$  (c)

### Problem 4

$$E_1 = \frac{\hbar^2 \pi^2}{2m_e L^2} = \frac{3.81 \text{ eV} \cdot \text{\AA}^2 \cdot \pi^2}{25 \text{\AA}^2} = \boxed{1.50 \text{ eV}} \quad (a)$$

(b) Inside the barrier, the electron wavefunction is

$$\Psi \sim e^{-\alpha x} = e^{-x/\delta}$$

$$\text{with } \alpha = \sqrt{\frac{2m_e}{\hbar^2} (U-E)} = \sqrt{\frac{1}{3.81 \text{ eV} \cdot \text{\AA}^2} (5.0 \text{ eV} - 1.5 \text{ eV})} = \frac{3.57}{\text{\AA}}$$

$$\Rightarrow \delta = 1/\alpha = 0.28 \text{\AA}$$

So the effective width of the well is  $\sim 5 \text{\AA} + \delta = 5.28 \text{\AA}$ .

So a better estimate of the ground state energy is

$$E_1' = \frac{\hbar^2 \pi^2}{2m_e (L+\delta)^2} = 1.50 \text{ eV} \cdot \frac{L^2}{(L+\delta)^2} = 1.50 \text{ eV} \left( \frac{5}{5.28} \right)^2 = 1.35 \text{ eV}$$

$$\Rightarrow \boxed{E_1' = 1.35 \text{ eV}}$$

(c)

$$T = e^{-2\alpha L} = e^{-2 \times 3.57 \text{\AA}^{-1} \times 1 \text{\AA}} = e^{-7.14} = 7.9 \times 10^{-4}$$

(d) Momentum  $p = m_e v = \hbar k = \frac{\hbar \pi}{L} \Rightarrow v = \frac{\hbar \pi}{m_e L} \Rightarrow$

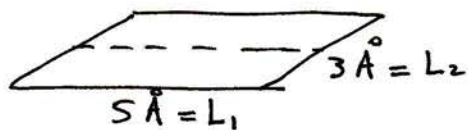
$$\frac{v}{c} = \frac{\hbar c \pi}{m_e c^2 L} = \frac{1973 \text{ eV} \cdot \text{\AA} \cdot \pi}{0.511 \cdot 10^6 \text{ eV} \cdot 5 \text{\AA}} = \boxed{0.0024} \Rightarrow v = \boxed{7.3 \times 10^{15} \frac{\text{\AA}}{\text{s}}}$$

(e) The electron travels  $2L = 10 \text{\AA}$  every time it hits the right well.

$$\text{So in 1s it hits } \frac{v}{2L} \text{ times} = \boxed{7.3 \times 10^{14} \text{ times}}$$

(f) If every time it hits barrier the probability of tunnel out is  $T = 7.9 \times 10^{-4}$   
the probability it will tunnel out in a 1s time interval is  $\sim 1$

# Problem 5



$$(a) \quad E_{n_1, n_2} = \frac{\hbar^2 \pi^2}{2m_e} \left( \frac{n_1^2}{L_1^2} + \frac{n_2^2}{L_2^2} \right) = \frac{\hbar^2 \pi^2}{2m_e L_1^2} \left( n_1^2 + n_2^2 \frac{L_1^2}{L_2^2} \right) =$$

$$= 1.50 \text{ eV} (n_1^2 + n_2^2 \times 2.778)$$

$$\Rightarrow E_{11} = 5.67 \text{ eV}, \quad E_{21} = 10.17 \text{ eV}, \quad E_{12} = 18.17 \text{ eV}, \quad E_{31} = 17.66 \text{ eV}$$

So the 3 lowest levels are:

$$n_1, n_2 = 1, 1 \quad ; \quad E_{11} = 5.76 \text{ eV}$$

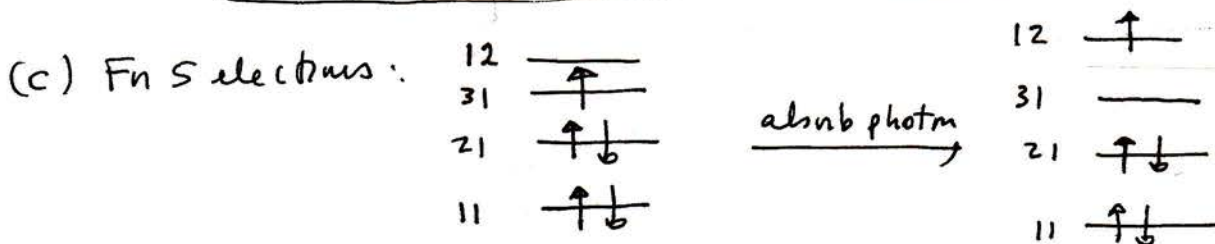
$$n_1, n_2 = 2, 1 \quad ; \quad E_{21} = 10.17 \text{ eV}$$

$$n_1, n_2 = 3, 1 \quad ; \quad E_{31} = 17.66 \text{ eV}$$

(b) To have  $\Psi(x, y = L_2/2) = 0$  we need  $n_2 = 2$ , so lowest state is

$$(n_1, n_2) = (1, 2) \quad ; \quad E_{12} = 18.17 \text{ eV.} \quad \text{Wavefunction is}$$

$$\Psi(x, y) = \sqrt{\frac{2}{L_1}} \sqrt{\frac{2}{L_2}} \sin \frac{\pi x}{L_1} \sin \frac{2\pi y}{L_2}$$



$$E = 2E_{11} + 2E_{21} + E_{31} = 49.34 \text{ eV}$$

the lowest excitation is electron in B1  $\rightarrow$  12  $\Rightarrow$

$$\Delta E = 18.17 \text{ eV} - 17.66 \text{ eV} = 0.51 \text{ eV} = \frac{hc}{\lambda}$$

$$\Rightarrow \lambda = \frac{hc}{\Delta E} = 24,314 \text{ \AA}$$

### Problem 6

$$\Psi(r, \theta, \phi) = C r^2 e^{-r/3a_0} \sin^2 \theta e^{i\phi}$$

(a) since exponential part is  $e^{-r/na_0} \Rightarrow \boxed{n=3}$

Since  $R(r)$  has no nodes,  $l = n - 1 \Rightarrow \boxed{l=2}$

Since azimuthal part is  $e^{im_e\phi} \Rightarrow \boxed{m_e=1}$

(b)  $P(r) = r^2 R^2(r) = C^2 r^6 e^{-2r/3a_0}$

$$P'(r) \propto 6r^5 - \frac{2}{3a_0} r^6 = 0 \Rightarrow r = \frac{6 \cdot 3a_0}{2} \Rightarrow \boxed{r = 9a_0}$$

In the Bohr atom,  $r_n = n^2 a_0$ ,  $n=3 \Rightarrow \boxed{r_3 = 9a_0}$  same

(c) Normalization:

$$1 = \int_0^\infty dr P(r) = C^2 \int_0^\infty dr r^6 e^{-2r/3a_0} \Rightarrow \boxed{C^2 = \frac{1}{\int_0^\infty dr r^6 e^{-2r/3a_0}}}$$

$$\langle r \rangle = \int_0^\infty dr r P(r) = C^2 \int_0^\infty dr r^7 e^{-2r/3a_0} = \frac{\int_0^\infty dr r^7 e^{-2r/3a_0}}{\int_0^\infty dr r^6 e^{-2r/3a_0}}$$

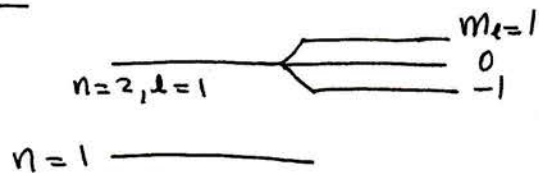
Using  $\int_0^\infty dr r^n e^{-\lambda r} = \frac{n!}{\lambda^{n+1}} \Rightarrow$

$$\langle r \rangle = \frac{7!}{\left(\frac{2}{3a_0}\right)^8} \cdot \frac{\left(\frac{2}{3a_0}\right)^7}{6!} = \frac{7}{\frac{2}{3a_0}} = \frac{21}{2} a_0 = \boxed{10.5 a_0}$$

(d)  $\langle \frac{1}{r} \rangle = \int_0^\infty dr \frac{1}{r} P(r) = \frac{5!}{\left(\frac{2}{3a_0}\right)^6} \cdot \frac{\left(\frac{2}{3a_0}\right)^7}{6!} = \frac{1}{6} \cdot \frac{2}{3a_0} = \boxed{\frac{1}{9a_0}}$

In the Bohr atom,  $\boxed{\frac{1}{r_n} = \frac{1}{n^2 a_0} = \frac{1}{9a_0} \Rightarrow \text{same}}$

## Problem 7



in magnetic field:  $U = -\vec{\mu} \cdot \vec{B} = -\mu_z B$

without spin,  $\mu_z = -\mu_B m_l \Rightarrow U = +\mu_B B m_l$

So:  $E_2 \rightarrow E_2 + \mu_B B m_l$

$$\mu_B B m_l = 5.79 \times 10^{-5} \text{ eV} \cdot 15 m_l = \boxed{8.69 \times 10^{-4} m_l \cdot \text{eV}}$$

So energies of photons emitted in the transition from  $n=2$  to  $n=1$  are

$$\boxed{h f = E_2 - E_1 \pm 8.69 \times 10^{-4} \text{ (} m_l = \pm 1 \text{) and } E_2 - E_1 \text{ (} m_l = 0 \text{)}}$$

with  $E_2 - E_1 = 13.6 \text{ eV} \times \frac{3}{4} = 10.2 \text{ eV}$

(b) With spin:  $\mu_z = -\frac{e}{2m_e} (\hbar m_l + 2\hbar m_s) = -\mu_B (m_l + 2m_s)$

$$U = -\mu_z B = \mu_B B (m_l + 2m_s) = \mu_B B \begin{pmatrix} 2 \\ 1 \\ 0 \\ -1 \\ -2 \end{pmatrix} \begin{matrix} m_l = 1, m_s = \frac{1}{2} \\ m_l = 0, m_s = \frac{1}{2} \\ m_l = 1, m_s = -\frac{1}{2} / m_l = -1, m_s = \frac{1}{2} \\ m_l = 0, m_s = -\frac{1}{2} \\ m_l = -1, m_s = -\frac{1}{2} \end{matrix}$$

(c) The difference in energy between the  $m_l = -1$  and  $m_l = 0$  states is  $8.69 \times 10^{-4} \text{ eV}$

At low temperatures, most electrons will be in lowest energy state  $m_l = -1$ .

Since the relative probability is given by the Boltzmann factor

$$e^{-(E(m_l=0) - E(m_l=-1)) / k_B T}, \text{ this becomes small when}$$

$$k_B T < E(m_l=0) - E(m_l=-1) \Rightarrow \boxed{T < \frac{8.69 \times 10^{-4} \text{ eV}}{\frac{1}{11,600} \frac{\text{eV}}{\text{K}}} = 10.1 \text{ K}}$$

### Problem 8

$$\Psi(x) = C \cos(kx), \quad [p] = \frac{\hbar}{i} \frac{d}{dx}$$

(a)  $Q$  is a sharp observable if  $[Q]\Psi = q\Psi$ , with  $q$  a number.

$$[p]\Psi = \frac{\hbar}{i} \frac{d}{dx} C \cos(kx) = -\frac{\hbar k}{i} C \sin(kx) \neq q \cdot C \cos(kx)$$

$\Rightarrow$   $p$  is not a sharp observable

(b)  $[p^2] = \frac{\hbar}{i} \frac{d}{dx} \frac{\hbar}{i} \frac{d}{dx} = -\hbar^2 \frac{d^2}{dx^2}$

$$[p^2]\Psi = -\hbar^2 \frac{d^2}{dx^2} C \cos(kx) = \hbar^2 k^2 C \cos(kx) = q\Psi$$

with  $q = \hbar^2 k^2$  a number. So  $p^2$  is a sharp observable

(c) Since  $p^2$  is sharp, there is no uncertainty,  $\Delta(p^2) = 0$ .

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

$\langle p \rangle = 0$  because wave function is even:  $\int_{-a}^a dx \cos(kx) \frac{\hbar}{i} \frac{d}{dx} \cos(kx) = 0$

$\langle p^2 \rangle = \hbar^2 k^2 \int dx |\Psi(x)|^2 = \hbar^2 k^2 \cdot 1$  since  $\Psi$  is normalized

$\Rightarrow$   $\Delta p = \hbar k$