

Problem 1

Assuming $V_0 \gg E$ (check later), the energy of this electron is

$$E = \frac{\hbar^2 \pi^2}{2mL_1^2} = \frac{3.81 \pi^2}{25} \text{ eV} = 1.50 \text{ eV}$$

The speed of this electron is:

$$v = \frac{p}{m} = \frac{\hbar k}{m} = \frac{\hbar \pi}{mL_1} \Rightarrow \frac{v}{c} = \frac{\hbar c \pi}{mc^2 L_1} = 0.0024$$

$$\Rightarrow v = 0.0024c = 7.27 \times 10^{14} \frac{\text{\AA}}{\text{s}}$$

the probability that the electron will tunnel through the barrier when it hits it is $T = e^{-2S\Delta x}$, $S = \sqrt{\frac{2m}{\hbar^2}(V_0 - E)}$, $\Delta x = L_2 - L_1 = 3 \text{\AA}$

The time it takes the electron to go back and forth once is

$$t = \frac{2L_1}{v}$$

\Rightarrow the number of times per second it tries to tunnel is $N = \frac{1}{t} = \frac{v}{2L_1}$

The mean lifetime τ is given by $\frac{1}{\tau} = NT$

$$\text{We have } \tau = 1 \text{ s} \Rightarrow T = \frac{1}{\tau} = \frac{2 \times 5}{7.27 \times 10^{14}} = 1.38 \times 10^{-14}$$

$$\Rightarrow -2 \sqrt{\frac{2m}{\hbar^2}(V_0 - E)} \Delta x = -15.96 \Rightarrow V_0 - E = 107.81 \text{ eV} \Rightarrow$$

$$\boxed{V_0 = 109.31 \text{ eV}} \quad \text{since } V_0 \gg E, \text{ infinite well approximation is good}$$

$$(b) \text{ For } \tau = 10 \text{ s} \Rightarrow \frac{e^{2S\Delta x'}}{e^{2S\Delta x}} = 10 \Rightarrow 2S(L_2' - L_2) = \ln 10 \Rightarrow$$

$$L_2' = L_2 + \frac{\ln 10}{2S}, \quad S = 5.32 \text{ \AA}^{-1} \Rightarrow \boxed{L_2' = 8.22 \text{ \AA}}$$

Problem 2

$$(a) \quad \Psi(r, \theta, \phi) = C r^3 e^{-r/a_0} f(\theta) g(\phi)$$

Exponential part of radial wavefunction is always $e^{-zr/n a_0} \Rightarrow$

for this wavefunction $\boxed{n = z}$

$$\text{Energy is } E = -E_0 \frac{z^2}{n^2} \Rightarrow \boxed{E = -E_0 = -13.6 \text{ eV}}$$

(b) From the form of the radial wavefunction $\Rightarrow n = 4$ (r^3 factor)

$$\Rightarrow \boxed{z = 4}$$

(c) From the form of the radial wavefunction (no nodes) $\Rightarrow l$ has maximum value $\Rightarrow l = 3$. The possible m values are $-3, -2, -1, 0, 1, 2, 3$

(d) Shortest wavelength photon it can emit: selection rule $\Delta l = \pm 1 \Rightarrow$

has to go to $n = 3, l = 2$ state \Rightarrow

$$\Delta E = \frac{hc}{\lambda} = E_0 z^2 \left(\frac{1}{9} - \frac{1}{16} \right) = E_0 \cdot \frac{7}{9} \Rightarrow \lambda = \frac{9}{7} \frac{12,400}{13.6} \text{ \AA}$$

$$\Rightarrow \boxed{\lambda = 1172 \text{ \AA}}$$

Problem 3

$$P(r) = C r^2 R^2(r) = C r^8 e^{-2r/a_0}$$

Most probable r : $P'(r) = 0 \Rightarrow$

$$8r^7 - \frac{2r^8}{a_0} = 0 \Rightarrow \boxed{r = 4a_0 \equiv \Gamma_m}$$

Bohr orbit for $n=4, Z=4$:

$$\Gamma_n = n^2 \frac{a_0}{Z} = 4a_0 \Rightarrow \text{same as most probable } r$$

$$(b) \quad \left\langle \frac{1}{r} \right\rangle = \frac{\int_0^\infty dr r^7 e^{-2r/a_0}}{\int_0^\infty dr r^8 e^{-2r/a_0}} = \frac{7! a_0^8 2^9}{2^8 8! a_0^9} = \frac{2}{8a_0} = \frac{1}{4a_0}$$

$$\Rightarrow \boxed{\left\langle \frac{1}{r} \right\rangle = \frac{1}{4a_0} = \frac{1}{\Gamma_m}}$$

$$(c) \quad \langle r \rangle = \frac{\int_0^\infty dr r^9 e^{-2r/a_0}}{\int_0^\infty dr r^8 e^{-2r/a_0}} = \frac{9! a_0^{10} 2^9}{2^{10} 8! a_0^9} = \frac{9}{2} a_0$$

$$\Rightarrow \boxed{\langle r \rangle = \frac{9}{2} a_0}$$

$$\text{or } \boxed{\langle r \rangle = 1.125 \Gamma_m}$$