

Problem 1

(a) The 10g of ice occupy volume $10 \text{ cm}^3 = 0.01 \text{ L} \Rightarrow$ we can ignore it compared with 1.5 L.

1 mol of air occupies 22.4 L \Rightarrow we have $n = \frac{1.5}{22.4} = 0.067 \text{ mols of air}$

molecular ~~atome~~ weight = 29 \Rightarrow 1.94 g of air

(b) Relative humidity is 100% because lid is sealing it

(c) Heating up to $100^\circ\text{C} = 373 \text{ K}$ we have

$$P_{\text{air}} = P_{\text{air}}(0^\circ\text{C}) \times \frac{373}{273} = 1.37 \text{ atm}$$

The water vapor has saturated vapor pressure = 1 atm \Rightarrow

$$P_{\text{total}} = 2.37 \text{ atm}$$

(d) We have water vapor at pressure 1 atm, volume 1.5 L, $T = 373 \text{ K}$

$$\Rightarrow n_{\text{vapor}} = \frac{1.5}{22.4} \times \frac{273}{373} = 0.049 \text{ mol}$$

molecular weight = 18 \Rightarrow $0.88 \text{ g of water vapor}$

\Rightarrow $9.12 \text{ g of liquid water}$

Problem 2

$$v_{rms} = 600 \text{ m/s} ; \bar{v}^2 = v_{rms}^2$$

$$E_{kin} = \frac{1}{2} m \bar{v}^2 = \frac{3}{2} kT \Rightarrow T = \frac{m \bar{v}^2}{3k}$$

$$\Rightarrow T = \frac{20 \times 1.66 \times 10^{-27} \times 600^2 \text{ K}}{3 \cdot 1.38 \times 10^{-23}} \Rightarrow \boxed{T = 289 \text{ K}} \quad (a)$$

$$(b) \quad g(v) = C v^2 e^{-\frac{1}{2} \frac{m v^2}{kT}}$$

$$\text{use that } \boxed{\frac{m}{kT} = \frac{3}{v_{rms}^2}}$$

$$\frac{g(v_1)}{g(v_2)} = \frac{v_1^2}{v_2^2} e^{-\frac{3}{2 v_{rms}^2} (v_1^2 - v_2^2)}$$

$$v_1 = 1200 \text{ m/s} = 2 v_{rms}$$

$$v_2 = 600 \text{ m/s} = v_{rms}$$

$$\Rightarrow \frac{3}{2} \frac{(v_1^2 - v_2^2)}{v_{rms}^2} = \frac{3}{2} (4 - 1) = \frac{9}{2} \Rightarrow$$

$$\frac{g(v_1)}{g(v_2)} = 4 e^{-9/2} = 0.0444. \text{ Now the interval } dv_1 = 10 \text{ m/s,}$$

$$dv_2 = 5 \text{ m/s} \Rightarrow \boxed{89 \text{ molecules with } v \text{ between } 1200 \text{ m/s and } 1210 \text{ m/s}}$$

$$(c) \quad f(v_x) = C e^{-\frac{m}{2kT} v_x^2}$$

$$\frac{f(v_{1x})}{f(v_{2x})} = \frac{f(0)}{f(v_{rms})} = e^{\frac{m}{2kT} v_{rms}^2} = e^{1.5} = 4.482$$

$$\Rightarrow \boxed{4,482 \text{ molecules with } v_x \text{ between } -1 \text{ m/s and } 1 \text{ m/s}}$$

(d) The force exerted in 1 collision is

$$F_i = \frac{\Delta(mv)}{\Delta t} = \frac{2mU_x}{\Delta t} ; \Delta t = \frac{2l}{U_x}, \text{ with } l \text{ the length}$$

of the container in that direction $\Rightarrow F_i = \frac{mU_x^2}{l}$

We have 1 mol = 6.02×10^{23} molecules, force = 9000 N \Rightarrow

$$l = \frac{N_A m U_x^2}{9000 \text{ N}}$$

$$mU_x^2 = kT = 1.38 \times 10^{-23} \times 289 \text{ J} \Rightarrow$$

$$l = \frac{6.02 \times 10^{23} \times 1.38 \times 10^{-23} \times 289}{9000} \text{ m} = 0.267 \text{ m} = \boxed{27 \text{ cm}}$$

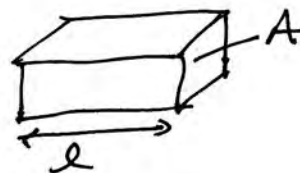
The time between 2 collisions is $\Delta t = \frac{2l}{U_x}$

estimate $U_x \sim U_{\text{rms}}$,

$$\Delta t = \frac{0.267 \text{ s}}{600} = 0.00045 \text{ s} = 0.4 \text{ ms}$$

Alternative derivation:

$$F = P \cdot A = \frac{N_A kT}{V} \cdot A = \frac{N_A kT}{l}$$



$$\Rightarrow l = \frac{N_A kT}{F} = 27 \text{ cm}$$

Problem 3

$$\rho = \frac{m}{V}, \quad \frac{\Delta \rho}{\rho} = - \frac{\Delta V}{V}$$

$$\frac{\Delta V}{V} = 3\alpha \Delta T = 3 \times 12 \times 10^{-6} \times 40 = 1.44 \times 10^{-3}$$

$$\Rightarrow \Delta \rho = -1.44 \times 10^{-3} \times 2,300 \frac{\text{kg}}{\text{m}^3} = -3.3 \frac{\text{kg}}{\text{m}^3}$$

$$\Rightarrow \rho(50^\circ\text{C}) = 2,297 \text{ kg/m}^3$$

$$(b) \quad \frac{F}{A} = E \cdot \frac{\Delta l}{l}; \quad \frac{\Delta l}{l} = \alpha \Delta T = 4.8 \times 10^{-4}$$

$$\Rightarrow \frac{F}{A} = E \alpha \Delta T = 20 \times 10^9 \times 12 \times 10^{-6} \times 40 \frac{\text{N}}{\text{m}^2} = 9.6 \times 10^6 \text{ N}$$

(c) For the F/A to be $3 \times 10^6 \text{ N/m}^2$, we need

$$\frac{\Delta l}{l} = 12 \times 10^{-6} \times 40 \times \frac{3}{9.6} = 1.5 \times 10^{-4}$$

so we need a spacer with $\frac{\Delta l}{l} = (4.8 - 1.5) \times 10^{-4} = 3.3 \times 10^{-4}$

$$\text{with } l = 3 \text{ m} \Rightarrow \Delta l = 0.99 \text{ mm} \sim 1 \text{ mm space between blocks}$$