

**PHYSICS 140A : STATISTICAL PHYSICS**  
**HW ASSIGNMENT #3**

**(1)** The entropy for a peculiar thermodynamic system has the form

$$S(E, V, N) = Nk_B \left\{ \left( \frac{E}{N\varepsilon_0} \right)^{1/3} + \left( \frac{V}{Nv_0} \right)^{1/2} \right\},$$

where  $\varepsilon_0$  and  $v_0$  are constants with dimensions of energy and volume, respectively.

(a) Find the equation of state  $p = p(T, V, N)$ .

(b) Find the work done along an isotherm in the  $(V, p)$  plane between points A and B in terms of the temperature  $T$ , the number of particles  $N$ , and the pressures  $p_A$  and  $p_B$ .

(c) Find  $\mu(T, p)$ .

**(2)** The Dieterici equation of state is

$$p(v - b) = RT e^{-a/vRT},$$

with  $v$  the molar volume and with  $a$  and  $b$  constants.

(a) What are the dimensions of  $a$  and  $b$ ?

(b) Find the coefficient of isobaric volume expansion,  $\alpha_p = v^{-1}(\partial v / \partial T)_p$ .

(c) Find the conditions for the inversion temperature of throttling,  $T\alpha_p = 1$  in terms of  $T$  and  $v$ .

(d) Define the temperature and pressure scales  $RT_0 \equiv 2a/b$  and  $p_0 \equiv 2a/b^2$ . Define also the dimensionless temperature  $\tau \equiv T/T_0$  and dimensionless pressure  $\pi \equiv p/p_0$ . Find and sketch the inversion curve  $\pi(\tau)$ .

**(3)** Consider the analog of the van der Waals equation of state for a gas of diatomic particles with *repulsive* long-ranged interactions,

$$p = \frac{RT}{v - b} + \frac{a}{v^2},$$

where  $v$  is the molar volume.

(a) Find the molar energy  $\varepsilon(T, v)$ .

(b) Find the coefficient of volume expansion  $\alpha_p = v^{-1}(\partial v / \partial T)_p$  as a function of  $v$  and  $T$ .

(c) Find the adiabatic equation of state in terms of  $v$  and  $T$ . If at temperature  $T_1$  a volume  $v_1 = 3b$  of particles undergoes reversible adiabatic expansion to a volume  $v_2 = 5b$ , what is the final temperature  $T_2$ ?